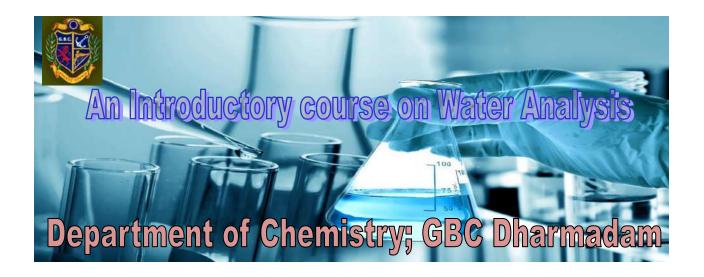


Value Added Courses



Government Brennen College

Dharmadam, Thalassery brennencollege@gmail.com brennenclg.dce@kerala.gov.in



Water analyses identify and quantify the chemical components and properties of water samples used in industrial processes, on waste-water stream, on rivers and stream, on rainfall and on the sea. The course offered by Department of Chemistry caters to the needs of the public for knowing the various water quality parameters, the importance of the water quality parameters, water pollution and the various techniques used in water analysis including BOD, COD, pH, TDS and DO.

Eligibility for the Course: A student of GovtBrennen College BA/BSc/MA/MSc.

GBC14VAC01: An Introductory Course on Water Analysis (2014 July 15 to 2014 November 30)

Syllabus:

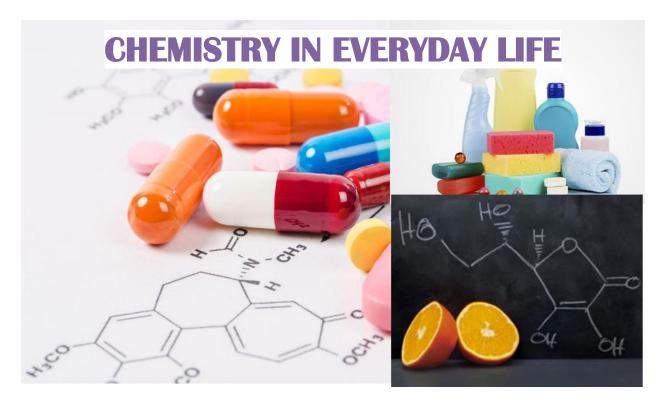
Unit 1: Introduction to Water

Unit 2: Water resources—Types of Ecosystems

Unit 3: Water Quality- Water quality parameters and their interaction- physical and chemical characteristics - turbidity, color - temperature - chemical constituents, taste, color, acidity, alkalinity - CO₂, hardness, pH

Unit 4: Methods of testing – Sampling

Unit 5: Practical Training on water analysis



Chemistry is the branch of science which deals with the investigation of the properties and changes of matter. From the way how our body exchanges oxygen to how our universe was created, all have a side of chemistry associated with it.Let us explore some common examples of chemistry in everyday life which most of us never knew about.

Eligibility for the Course: A student of Govt. Brennen College (BA/BSc/MA/MSc)

GBC15VAC01: CHEMISTRY IN EVERYDAY LIFE (With effect from 2015 December)

Aim: To create interest in studying the role of chemistry in overall development Objectives:

- 1. To provide knowledge about the structure, properties and uses of plastics &polymers
- 2. To provide a brief idea about the role of chemistry in industry and agriculture
- 3. To make aware about the use and abuse of chemicals in daily life
- 4. To study about the chemistry of medicines
- 5. To provide knowledge about adverse effects in using polluted water and methods of treatment

Syllabus:

Unit 1: PLASTICS & POLYMERS

Unit 2: FERTILIZERS & INSECTICIDES

Unit 3: FUELS, CELLS & BATTERIES

Unit 4: CEMENT &GLASS

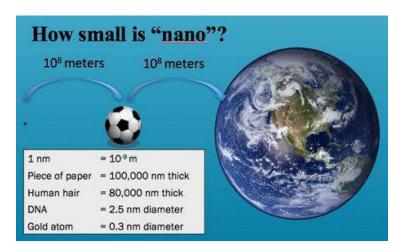
Unit 5: COSMETICS

Unit 6: MEDICINES

Participants will get a certificate after the completion of the course.

Nanoscience and Nanotechnology Department of Chemistry; GBC Dharmadam

Nanotechnology is the study and synthesis of materials in the nanodimension. Being the hot topic of the era, Chemistry department offers a course in Nanosciences and Nanotechnology this year. The course covers various aspects of nanoscience from historic perspectives to state of the art research in nanotechnology. The course introduces a general awareness about the nanotechnology, what it is about, the various synthetic pathways, the applications etc. The course would be extremely useful for students of Chemistry, Physics, Botany and Zoology students in specific and others in creating general awareness.



GBC16VAC01: Nanoscience and Nanotechnology (With effect from 2016 July)

Syllabus:

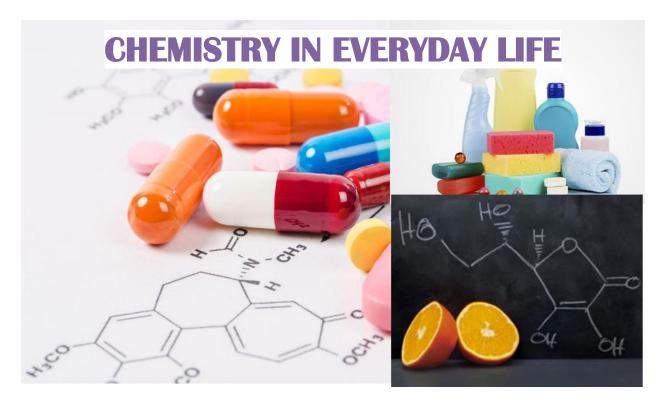
Unit 1: Introduction to Nano

Unit 2: Synthetic Approaches to nanofabrication

Unit 3: Optical, Magnetic, Biological Properties of Nano

Unit 4: Quantum Confinement, Tunneling

Unit 5: Applications of nanoparticles in Chemistry, Physics, Botany and Zoology



Chemistry is the branch of science which deals with the investigation of the properties and changes of matter. From the way how our body exchanges oxygen to how our universe was created, all have a side of chemistry associated with it.Let us explore some common examples of chemistry in everyday life which most of us never knew about.

Eligibility for the Course: A student of Govt. Brennen College (BA/BSc/MA/MSc)

GBC15VAC01: CHEMISTRY IN EVERYDAY LIFE (With effect from 2015 December)

Aim: To create interest in studying the role of chemistry in overall development Objectives:

- 1. To provide knowledge about the structure, properties and uses of plastics &polymers
- 2. To provide a brief idea about the role of chemistry in industry and agriculture
- 3. To make aware about the use and abuse of chemicals in daily life
- 4. To study about the chemistry of medicines
- 5. To provide knowledge about adverse effects in using polluted water and methods of treatment

Syllabus:

Unit 1: PLASTICS & POLYMERS

Unit 2: FERTILIZERS & INSECTICIDES

Unit 3: FUELS, CELLS & BATTERIES

Unit 4: CEMENT &GLASS

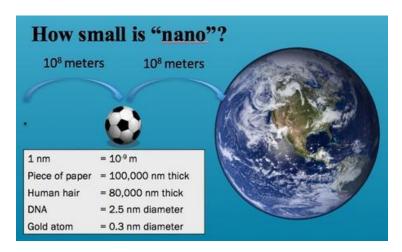
Unit 5: COSMETICS

Unit 6: MEDICINES

Participants will get a certificate after the completion of the course.

How small is small?? Department of Chemistry; GBC Dharmadam

Nanotechnology is the study and synthesis of materials in the nanodimension. Being the hot topic of the era, Chemistry department offers a course in Nanosciences and Nanotechnology this year. The course covers various aspects of nanoscience from historic perspectives to state of the art research in nanotechnology. The course introduces a general awareness about the nanotechnology, what it is about, the various synthetic pathways, the applications etc. The course would be extremely useful for students of Chemistry, Physics, Botany and Zoology students in specific and others in creating general awareness.



GBC18VAC01: Nanoscience and Nanotechnology (With effect from 2018 July)

Syllabus:

Unit 1: Introduction to Nano

Unit 2: Synthetic Approaches to nanofabrication

Unit 3: Optical, Magnetic, Biological Properties of Nano

Unit 4: Quantum Confinement, Tunneling

Unit 5: Applications of nanoparticles in Chemistry, Physics, Botany and Zoology



Green chemistry is the utilization of a set of principles to reduce or eliminate the use or generation of hazardous substances in the design, manufacture and applications of chemical products. The course offered by the department on "GREEN CHEMISTRY AND SUSTAINABILITY" familiarizes the students with environmental, economic, and societal benefits of green chemistry

Eligibility for the Course: A student of Govt. Brennen College BA/BSc/MA/MSc.

GBC14VAC02: GREEN CHEMISTRY AND SUSTAINABILITY (With effect from 2014December)

Syllabus:

Unit 1: Importance of environmental protection

Unit 2: Principles of green chemistry

Unit 3: Toxicology and Green Chemistry

Unit 4: Green protocols in undergraduate laboratory classes

Unit 5: Energy and environment

GENERAL INTRODUCTION TO THE CHEMISTRY OF DYES

Department of Chemistry; GBC Dharmadam

Color and its various uses come over the horizon from pre-historical period by all cultural groups and on all major land masses. The chemistry behind the various colours is fascinating to man and the study of the materials leading to beautiful colours is of importance. The course on 'Chemistry of dyes' offers students to have an acquaintance with the fascinating world of colours, the natural and synthetic dyes, the chemistry behind colour, introduction to synthesis of dyes, practical session in dye synthesis and dyeing.

Eligibility for the Course: A student of GovtBrennen College BA/BSc/MA/MSc.

GBC15VAC02: GENERAL INTRODUCTION TO THE CHEMISTRY OF DYES (With effect from 2015 December)

Syllabus:

Unit 1: History of Dyes, Relation with dyes and colour, Colour and Structure

Unit 2: Classification of dyes

Unit 3: Toxicity of dyes, Medicinal Properties of dyes

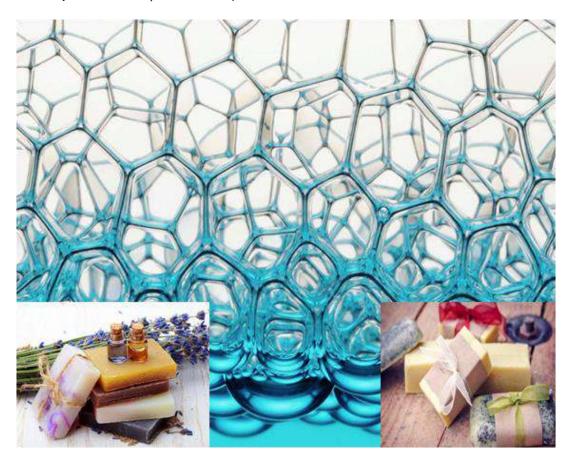
Unit 4: Basics of natural dyeing – Pretreatment, mordanting, preparation

Unit 5: Vat dyes, mordant dyes, anthraquinone, anthocyanine, flavonoid, carotenoid dyes

Unit 6: Practical application in dye synthesis and dyeing

HANDS ON TRAINING IN SOAP& DETERGENT MAKING

GBC16VAC02: (From 2016 July to Nov 2016)



Soaps and detergents are well known cleansers using in everyday life. They have the ability to remove dirt from human skin and textiles. Soap is a salt of long chain fatty acid. Soap consists of a long hydrocarbon chain with a carboxylic acid group on one end which is ionic bonded to a metal ion, usually a sodium or potassium. The hydrocarbon end is nonpolar and is soluble in nonpolar substances and the ionic end is soluble in water. This training program will be useful for students to make soaps and detergents conveniently by readily available precursors.

Eligibility for the Course: A student of GovtBrennen College BA/BSc/MA/MSc.

Duration: 30 Hours

Course Starting on July 15, 2016 extends upto Nov 10, 2016

Syllabus

Unit I: Introductory theoretical remarks on soaps and detergents

Unit II: Description of raw materials used and the synthetic procedure employed for the manufacure of soap and detergents

Unit III: Practical training on the synthesis of soap and detergents

CHEMISTRY LABORATORY TECHNICIAN COURSE



GBC17VAC02: Course Duration: 30 Hrs

Chemistry department offers a course for the laboratory technician training this year for the students, staff and teachers of Govt. Brennen college starting from Dec 2017 to February 2018.

MODULE I 2 Hrs

Common Apparatus and Instruments

- a) Familiarization of Common Apparatuses and Instruments
- b) Maintenance of Apparatuses and Instruments

MODULE II 3 Hrs

Mixtures and their Compositions

- a) Mixtures
- b) Composition and Concentration Terms
- c) Preparation Standard Solutions

MODULE III 4 Hrs

Preparation of Laboratory Reagents

- a) Preparation of Acid and Alkali Solutions
- b) Preparation of Bench Reagents
- c) Preparation of Special Reagents
- d) Preparation of Indicators

Module IV 3 Hrs

Volumetric Analysis

- a) Introduction to volumetric Analysis
- b) Types of Volumetric Analysis
- c) Indictors for Volumetric Analysis

Module V 3Hrs

Gravimetric Analysis

- a) Introduction to Gravimetric Analysis
- b) Precipitation and Filtration Techniques
- c) Drying and Weighing of Precipitate

Module VI 3 Hrs

Mixture Analysis

- a) Preparation of salt mixtures
- b) Preparation of H2S using Kipp's Apparatus and its maintenance

Module VII 4 Hrs

Analysis and Preparation of Organic Compounds

- a) Preparation of reagents for Organic Analysis
- b) Preparation of reagents for Organic Preparations

Module VIII 3 Hrs

Physical Chemistry Experiments

- a) Common Instruments and apparatuses for Physical Chemistry experiments
- b) Maintenance of Hot air oven, Stirrers and water baths

Module IX 4 Hrs

Lab Safety and Maintenance

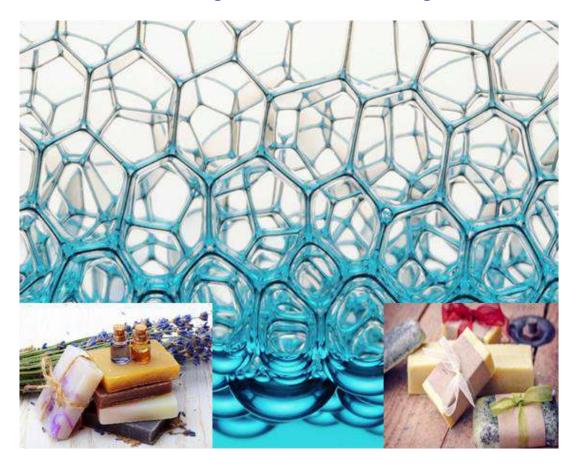
- a) Reading MSDS
- b) Introduction to Lab Safety Instruments
- c) Use of Lab Safety Instruments and their maintenance

Module X 2 Hrs

Laboratory Store Management

- a) Maintenance of Laboratory Store
- b) Maintenance of Stock Register

HANDS ON TRAINING IN SOAP& DETERGENT MAKING



Soaps and detergents are well known cleansers using in everyday life. They have the ability to remove dirt from human skin and textiles. Soap is a salt of long chain fatty acid. Soap consists of a long hydrocarbon chain with a carboxylic acid group on one end which is ionic bonded to a metal ion, usually a sodium or potassium. The hydrocarbon end is nonpolar and is soluble in nonpolar substances and the ionic end is soluble in water. This training program will be useful for students to make soaps and detergents conveniently by readily available precursors.

Eligibility for the Course: A student of GovtBrennen College BA/BSc/MA/MSc.

GBC18VAC02 - Soap Making and detergent making

Syllabus

Unit I: Introductory theoretical remarks on soaps and detergents

Unit II: Description of raw materials used and the synthetic procedure employed for the manufacure of soap and detergents

Unit III: Practical training on the synthesis of soap and detergents

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SL NO 05/ARB/18 07/ARB/18 06/ARB/18 04/ARB/18 03/ARB/18 01/ARB/18 09/ARB/18 08/ARB/18 15/ARB/18 14/ARB/18 13/ARB/18 12/ARB/18 11/ARB/18 10/ARB/18 19/ARB/18 18/ARB/18 ROLLN 29 27 26 64277 FATHIMA ABDUL LATHEEF 64554 MUHAMMED JASSAR S 64755 FAHEEMA V P 64514 ASHIFA ASHARAF V V 64405 MUHAMMED ASHRAF B 64759 MUHAMMED FAYIS K 64756 MUHAMMED FARHAN HABEEB M A 64279 HUSSAIN K K 64649 ARSHAD A V 66213 NAFEESATHUL MISIRIYA P P 64216 ZAMNA KHALID 64745 VAHEEDA T K 64486 SHAHSILA K T 63947 SARAH MOHAMED FAISAL 66407 NAJILA T V 66489 FADHIYA JEBIN U V 66523 ASEELA T M 66465 MUSHMILA N 66102 MOHAMMED WAJIV A C 65997 FATHIMATHUL SHAFA P 66345 FATHIMATHUL MUHSINA P 66502 FATHIMATHU ZANHA 66388 FATHIMATH NIDA D C 66485 FATHIMA IRFANA N P 66361 AYSHA SHEDHA K P 66041 AYISHATH FASNA K 66511 MUHAMMED SAHANAJ P K 66307 FIDA FATHIMA P 66060 FATHIMATHUL SINONA PARVEEN T 65961 FATHIMATHUL NEDHIYA 66473 FATHIMATH SAFFANA M K ADMN Signature of the Teacher GOVERNMENT COLLEGE DHARMADAM, THALASSERY ATTENDANCE REGISTER FOR THE SHORT TERM COURSE IN SPOKEN ARABIC LOWER NAME **DEPARTMENT OF ARABIC 2018-19** Z Z Z Z \leq Z M/F ZZ T , Li T T Dt T 19 20 3 3 3 P 3 32 33 34 35

2	3	2		DATE		1	1/5)	20	19	1 1	/12/	20	19	1/	26/	201	9	2/	9/20	019	1 2	/16	/201	19	2/	23/3	2019	3	3/2	/20	19	3	3/9/2	2019
3	ROLL	ADM	NAME	M/F /PE	RIGO	13	-	-	_	5	-	7	8	-	10	-	-	_	-	5 1	17	18	19										30	
1	66092	02/BOT/18	AKSHAY B R	MA		1×	×	×	v							. 1		x .		1	×	X	X	X		1	1	1.	1	1.		×	X	A
2	66146	03/BOT/18	AKSHAY V V	MA	E.	×	X	1 ×	X	X	x	X	X			-	-	N	N	XX	1.			•					1	1	VX	X	×	×
3	65814	04/BOT/18	ANAGHA K	FEM/	LE	×	×	×	X	X	N	X	8	X	X	x	1			17	1×	×	×	X	1	×	4/2	4	AF	X	X	X	X	×.
4	66488	05/BOT/18	JAYAPRAKASH K	FEMA	LE		-							X	A.	X	x		1	T	1.			•	7	7	X)	X >	1 >	42	(X	X	A	*
5	65821	06/BOT/18	ANUPRIYA N K	FEMA	LE	×	X	×	X		-	-	- 1	XI.	X,	X.	1	1/2	×	X	人	×	X	٨	X	X	٨)	()	* >	4 >	CX.		•	
6	65813	07/BOT/18	ANUSHA K.S	FEMA	LE	X	X	×	1	N	X	x	X 7	27	x /	9	(7	17	X	X	X	×	X	X					1.		1.	-		
7	66263	08/BOT/18	ARDRA RAMESH	FEMA	E	X	X	X	X	X	X	XI.	8/7	1/2	1	()	1			1	2	×	×	X	X	X	X	1 2	< >	AX	14	X	X	4
8	66260	09/BOT/18	ASWANIP	FEMAL	E			.		X	SI	0	41		1	1.	1	1	T		X	X	X	Y		1	1	17	YX	12	X	15	×	XV
9	66282	10/BOT/18	BHASKARAN	FEMAL	E	X	×L	×	XI.	X	XX	· x		1		1	17	NX	X	X	X	X	X	×	•		1.	17	VX	1	X	1 ×	X	X
10	65402	1/BOT/18	NAMBIAR	FEMAL	E	X.	x	4	1	1	1	7.	1×	Ts	X	1	1	VX	X	X	X	X	X	X	X :	× ;	S X	- 7	1/2	(X	X	1	7	41
1	66689 1	2/BOT/18	p i	FEMAL	T		1	1		1	1	1.	TX	1,	1)	1×	· x	X	X	X	1			-	X -	N)	1/2	1	1.	1.	11	-		
_			MITHUNE	MALE		1	1	1		1	x >	X	17	1	1	X	X	1	×	x			1	1	5 >	4 >	X	1		1		X	X	××
-	66120- 1	BOT/18 IN	VIMISHA K	FEMALE	1	XI)	10	Ix	1.	1.	1.	1.	1.	1.	1.	1.		1.		1	X	X	K X	1	4/1	1	X	14	X	×	×		•	,
	66247, 15	BOTHS S	HIFANOOR	FEMALE	15	do	1	Tx	/ x	13	12	dx	1	+		1	1.	-	-	4	X	x i	4	4	1.	,	1	1	1	1		1		
_			ISIRA SANIL	FEMALE	Tx	12	Tx	X	TX	1	2	Y	×	×	1 ×	14	17	1	~	×	X	人	1				-		1.			X	x	XX
_				FEMALE	1	1.	1	1	18	X	X	Y	10	A	X	K	×	X	X	V	지	×	X ;	6	1 7	Vy	1	7	X	1	X	×	×	X
-	and the same of			FEMALE	1	X	15	×	1	1	1	1	1				X	-2	X	×		1		. 1	人人	17	1	X	×	X	×	1	X	4,
-		OT/IS CA		FEMALE	1×	ź	12	10	1×	×	×	8			-		Ź	X	x	x	V 1	0	1	1	1.	1.	1.	×	×	×	1	X	X	X
	*	OT/18 IAN		EMALE	to	-	1x	12	12	×	×	2	*	*	x	+	1		-	7	2	*	X	1	1	1	12	1×	×	×	×	1	1	
	114 21/B		nigo and control of the control of t	EMALE	12	2	12	1	1.	-	-		X	A	X	y	-		-	-	N	2		1.	1.	1.	1.	t	X	×	20	-		
661				EMALE	1	-	1		-	-	-	-	1	(1)	-	1	×	~	2	x	1	1		1	1.	+	Ė	12	1	1	· ·	~	8	1
-				EMALE	1	^	1	1	2	*	20		+	-	÷	-	-	1	7			4	11	. 3	1 4	-	1	1	1	f.	1		-	
662			The state of the s		7	^	-3	X	\Rightarrow	9	싔	*	-	-	7	7	-	×	,	v	17	¥	*	12	10	10	10	1	1 3	-	-	2	2	-
	50 24/BC			MALE	X	^	^	X	A	4	4	2	24	4	4	4	A	4	4	1	1	1	+	+	1	1	X	1	1	1	A			
6630	-			IALE	1	×	X	1	XI.	1	4	4	4	+	-	•	Ă.	4	4	X		1	+	1	12	12	Y	1,				1	1	4
65850	·	-		MALE	1	.	1	1		1		1	-	1	*	1	7	4	×Ψ	()	7	1	1	1		1		1				X	A	4
6426	27/BOT	118 VALS	ARAJVN FE	MALE		1		1	1	1		1	X.	7	X	4				. 1	Y	V	X	1	1	*		X	1	X	X	X	X	7
5817	28/BOT	18 MALA	VIKAK FEN	MALE	XI	1	K/	1	X	KI.	XX	1	× ;	N.	7	2				. 3	X	×	X	17	X	X	X	大	X	X	X	X	×	X
5302	29/BOT/	IS MARIA	WAAT FEM	MALE	XI;	X.	XI>	11:	XX	()	11	1/2	4)	K/	× .	XI:	X)	X.	N)	(X	X	×	X		-	*		X	X	X
274	30/BOT/	8 R	FEM	ALE	1	T	1	T	1/2	()	< >	()	(1)	XX	· X		N)	N;	X)	1/2	4 ×	X	×	1	1	1	1	×	×	X	X	×	X	X
257	31/BOT/1	E IN	FEM	ALE	1/2	1	xx	17	JX	I×	1 1		T	-		17	()	1/1	1	1>	X	X	U		,		3	X	X	X	×	X	×	1
	32/BOT/1		PK FEM.	ALE 1	12	1,	ets	13	c ×	13	SI S		+	1		1	. 1	1	1		1	1.	1,	1	X	×		×	4	×	X	V	1	X
-	34/BOT/18				1/2	12	10	1-	1	12	12	12	17	X	1	1>	15	1/2		+	1	1		1	1	1	I.	1	1		1	1	×	2/2
inform.	and the same of the				10	10	10	15	1	15	12	15	to	10	12	15	15	1	10	1:	1:	1	15	×	×	×	×	-			1	12		~ >
	35/BOT/18				4	10	10	12	14	1	A	14	15	13	14	1	4	1	+	17	<u> </u>	X	^	1	10	^	A	-		-	-	1	7	-
0 3	6/BOT/18	VINAYA		1.0 17	1	1/	1/	1	10	X	1	1 7	1	1^	r	1.	1,	Τ.			_	_		_	_				ш		ш	ш	_	
			teacher																5	Sign	atur	e o	f the	e H	9	pro de	12	Ber	larti	OF!	PAST PAST W!	ITM HH	Hit I	LIGHT CA ME PI

			ATTENDANCE REGIS			GO	VER	NM	ENT	COL	RTM	ENT	OF	ВО	TAN	Y	HALA		lan t	-	22/0	9/2	017	(21	nd u	g 3r	d se	em)		_	0/0	/20:	17	\top	9/23	3/2
								CON	100	URS	EIN	GA	RDE	NIN	IG F	RON	124	/06,	201	7-1	13/0	8/	5/2	017	T	8,	19/	201	7	_	9/9	120.	7/2		9 30	
			ATTENDANCE REGIS	TER FOR T	HE:	HUI	(70	117	T	7/1	/20:	17	T	7/	8/20	17		7/2	2/2	5 1	-	ان و	el.	19	20	21	22	23	24	25	26	12	120	1	-	-
1	3 1	2		DAIL	-	-	-	31 4		5 6		7 1	8	9 1	0 1	1 1	2 1	3 1	-	-	0 1	4	1	,			×	*	X	X	X	×	-	+~	1	_
1	ROLL	ADM	NAME	M/F/PE		1 3	-	+	+	-	+	1	1	×		×	X			×μ	42	42	1	-	X	4	1	×	×	×	X	X	-	IX	-	-
	85		ADILA ABDULLA	FEMALE	-	14	X	1	÷	×	1	X	1	1				1	1	12	-	_	-	0	_	x l	V	-				K	\propto	×	X	1
1	86		FATHIMA T	FEMALE	X	×	X	×	12	-	1×	1 x	_	1	/ ×	1,	4 0	X O	10	(>	1	. 3	42	40	4	1	7	J	K	W	X	.x	12	1 ×	X	×
1	87		HAFIRA C K	FEMALE	7	×	X	×	×	×	X	-	-	x :	,	11	1 %	13	13	1 3	12	4	<u> </u>	4	X	7	3	S	×	×	×	×	X	X	X	X
5	89		ANIANA R	FEMALE	1	X	>	4 7	4	(×	-	12	_	7		1	2 1	T	1	· X	3	12	1	4	40	4	7		2	1	×	X	×	X	×	X
-	_		ANJANA VIJAYAN	FEMALE	12	X	1	K	()	47	N	X	-	+	1 3	1	1	/ x	X	1	10	4	X	V)	1		1	X	2		~	~	V	V	X	×
6	90		ANJUCHANDRAN P	FEMALE			1		12	42	X	X	42	4	13	-	HO	1/2	4	1~	· x	10	17	(1)	10	()	4	X	X	A	A.	-	13	2	7	V
7	91			FEMALE	d	JX	1 7	JX	12	X	X	X	TX	12	1	1	45	12	1	J	12	1-1	X	1)	(1)	2	1	X	X	X	X	X	×	1	x	÷
8	92		ANUSREE C ARUNIMA JAYADEV	FEMALE	13	N A	D	1	(k)	×	X	2	Y	12	4	-	-	1	1	15	10	×	X	×	13	X 6	(1:	X	X	X	X	A	X	\sim	X	2
9	93			FEMALE	1.		1.	1	×	X	13	X	X	1X	X	X	1	12	+	12	12	1	1	13	(×	(>	1					X	X	×	X	1
10	94		ARYA SUNIL V K	FEMALE	1×	x	X	x	X	X	12	X	X	1	4X	X	X	IA	1	1	10	13	13	1		. 12	KI O	X	X.	X	×	X	×	X	X	X
11	95	_	ATHEENA K	FEMALE		×	10	K	d	1.			1	X	X	×	X	12	X	13	×	+	1 ×	1	12	7.	1.	1	,	1	\propto	X	X	X	X	X
12	95	63841	DRISHYA V C	FEMALE	-	12	V	V	X	×	×	X	×	10X	X	1	X	1>	42	14	17	10	X	1	1	10	1/2	11	X	X	X					X
13	97		FATHIMATHUL MISILA A P	FEMALE	-	V	V	T _v	v	×	×	X	X	X	×	X	X	1	X	X	X	X	10	13	-	10	15	113	,				,	×	X	A
14	98		KAVYASARIGA N K	FEMALE	40	X	1	V	×	X	X	X	X	X	X	X	X	×	X	×	12	×	X	10	1	4	Y,	1)	X	1	1	X	x l	Y	4
15	99		KEERTHANA M	FEMALE	-	1.	1.		7	DX.	×	X	X	×	X	X	X	X	X	10	X	X	X	a	X	12	P	1	-	_	31	2	4/	2	-	~
16	100		NADIYA M K	FEMALE	+-	X	v	1~	1	1x	×	×	×	X	X	X	,	1	,	,	×	X	13	X	×	IX	10	-	X	10	1	7	-	4	1	×
17	101		NANDANA VUAY	FEMALE	12	-	1	1	1	L	m	x	X	X	X	×	1 4	X	2	X	ox	×	X	×	3	0	13	19	-	5 3	-		X		1	4
18	102	-	NIVYA MARIYA T	FEMALE	×	100	01	V	V	2	2	×	2	×	N	¥	X	X	x	X	X	X	X	X	X	2	4 ×	>	x 2	3	X 2	X	즸	+		+
19	103	-	SITHARA A P M	FEMALE		/X	Ť	1	-	-	1	-	×	V	×	1	X	×	×	×		+			X	1	X	X	12	1/	1	4	X	•	-	4
20	104		SRUTHI S NAIR	MALE	×	V	-	2	.,	1		-	V	×	X	×	×	×	×	×	×	X	×	X	1 %	X	1	V	1	· X	5 3	5	4	_	X >	-
21	105		ARUN B A	MALE	1	J.	7	J	Y	x	1		1	x	x	X	×	×	X	×	X	X	x	X	-			1.	×	10	(>	K X	0 3	K.	X >	×.
22	106		NEERAJ ANANDRAJ	FEMALE	X	d	7		10		X	OV.	×	A	A	×	又	X	×	×	×	×	X	x				1.	12	4 >	XX	1	1 1	1	x	~
23	107		ANAGHA M K	FEMALE	1	7	1	×		×	v	1	4	*	1	x	X	TX.	×	v	X	X	×	×		,		10	13	17		Z X				T
24	108		FATHIMATH JUMANA HASIN K B	FEMALE	1	A.	a.	^	X	V	V	X	×	×	2	*	X	X	×	×	×	×	-		×	×	×	X	17	1×	1	12	7	× /	47	1
25	109	_	JYOTHI KRISHNA S	-	X	4	X	X	X	4	1	3	-0	^	1	2	V	÷	-	X	-	-	^		×			5	×	1	1	1	-	-		-
26	110		MUNEEBA M K	FEMALE	3	1	7	X	h	ox.	×	ox I	7	-	7	1	~	-		2	×	x	x	,	×	×	K	X	×	×		-	1 3	* *	1 1	13
27	111		NAINU VIJAYAN	FEMALE	X	a	X	«	V	×	1	4		-	-	-	7	7	싓	×	9	3	-	A	*	-	_	1	1	1	10	X	+	\ X	-	×
28	112		SREESHA K S	FEMALE	N	4	Α,	V		×	×	N	-	÷	•	:	4	4	4	^	^	-		A		^	^	X	+÷	H	+	+	1	24	X	10
29	113		VISMAYA K T	FEMALE	4	3	X	N	X	-	¥	Y	4	×	4	4	^	7	×	× I		-		•	^	٨	×	×	-	Η.	+:	+.	15	1		10
30	114	-	VYSHNA T	FEMALE	X	A	X	X	14	od	X	X	3	1	1	X	•	•		•	X	X	×	V	^	×	^	×	×	X	X	IX	12	X	-	1
31	115	-	AKSHAY K M -	MALE	X	X	A	V	X	Y	X	15	N	4	4	X	X	X	*	X	4	4	4	×	7	^	X	A	×	X	×	X	X	1 1	X	X
32	116		ANANDAKRISHNAN M K	MALE	Х	X	A	X	Y	X	X	X	4	X	×	1				•	X		×	4	7	1	×	1	×	×	X	×	1	X	1	1
34	117		ARJUN K P BAUU K S	MALE	V	4	M	1	×	V	X	V.		-	-1	•	X	×	×	X	_	Α.	٨.	X	1	X	X	X	1	×	×	1	1.			
35	120		MITHUN.P.C	MALE	X.	X	X	X	X	a	X	X			•	•	٠	-		.	A	X.	X.	K					+	X	14	4	X	×	X	oc
36	121		RAHUL P P	MALE	X	X	X	X	X	N	4	X.	×	1	1	*				*	Χ.	^	× .	X	N	×	X	1	X	×	4	4	×	X	X	X
37	123		VISWAJITH V S	MALE	V.	X.	X	X	X	X	X	X,	^	1	_	4		_	× >	0		•		. 1	×	×	×	1	30				x	×	X	Y
38	124		ANJANA HAREENDRAN K V	MALE	X	X	2	D.	X	×	X.	X	Λ	X	^	x	× .	×,	4)	K	- [•							×	×	~	x				
100	124		gnature of the Teacher	FEMALE	×	a	X	1	X	X	X	*	•				-		. [1	9	1	x	지	ス	~	×	×	1	×	×	1			
_	_	3	grature of the feacher		-						_							\neg	1		T	1	1	1	+	+	1	$^{\sim}$	1	4		\sim	-	\vdash	-	$\dot{-}$
		740																		-	Sign	tur	e of	the	He	ad o	Fab	a D	epari	tma		_		_	_	_
						11										41	OVI-				0		U	44.14	, 10.	ou u		1	par	nue	WAT.					

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